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The oxidation number of a free element is always 0. The atoms in He and N₂, for example, have oxidation numbers of 0. The oxidation number of a monatomic ion equals the charge of the ion. For example, the oxidation number of Na⁺ is +1; the oxidation number of N³⁻ is -3. The usual oxidation number of hydrogen is +1.

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Ionic Bonding Chemistry If8766 Answers

Solution for Assign the correct oxidation number to each species in the given redox reaction. (Pool: 3 of 5) Assign the correct oxidation number to each species...

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