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## Chemistry 2nd Semester Study Guide Answer Key

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semester honors. chapter 14: mixtures and solutions. Note: Section 14.2 only pages 480-top of 482 are covered. Follow same format for independent work as explained for chapter 3 on the first semester page. ... study guide and answers. problems from class notes. supplementary problems and ...

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Chemistry Exam 2nd Semester. combination reaction. decomposition reaction. combustion reaction. single replacement reaction. a chemical change in which two or more substances react to form.... a reaction in which a single compound breaks down to form two.... a chemical reaction that occurs when a substance reacts with o....

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Chemistry - 2nd Semester Practice Test Thermochemistry (Ch.11): Multiple Choice 24) What is the amount of heat required to raise the temperature of 200.0 g of aluminum by  $10\Delta C$ ? cal (specific heat of aluminum = 0.21 ) g  $\Delta C$  24) A) 4200 cal B) 420 cal C) 42 000 cal D) 420 000 cal E) none of the above

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YHS Chemistry Second Semester Final Exam Study Guide 1. What mass of hydrogen gas can be produced by reacting 6.92 g of sodium in excess water? The reaction is: 2. Calculate the masses of silver chloride and dihydrogen monosulfide formed when 85.6 g of silver sulfide reacts with excess hydrochloric acid, HCl. (double replacement reaction) a. AgCl = 99.0g b. H<sub>2</sub>S = 11.8g 3.

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These guides are closely aligned with chapters 12-17 and 21 and were designed for use in the second semester sequence of general chemistry. They can be used in a flipped-style classroom where students complete them before the lecture. Or they can be used to reinforce important topics learned in class.

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